Weather and Climate • Two fundamental laws: Can use this to figure out • Wien's law: • Stefan- Boltzmann law how hot the earth should be · Wavelength of peak i.e. as we heat up objects, • Total Power radiated/unit area • Our model: both earth and sun are perfect they go black bodies • black \Rightarrow red \Rightarrow orange \Rightarrow yellow \Rightarrow white • Sun is 7x105 km in radius, temp. of 5800 K. How much energy does it radiate? • (remember: must work in absolute temp, K • How much is absorbed by earth? not °C) • What temp. would earth be at to re-radiate • i.e double the temp, 16 times the energy this? Peter Watson 810.47 1+10 1 2+10 absorbed by earth •Sun produces radiation Inflow from sun must Predicts Each square metre produces σT_{sun}^4 , so total power output balance outflow (luminosity) is $L = 4\pi R_{sun^2} \sigma T_{sun^4}$ Inflow Temp of Earth is This gets emitted into Outflow space, spreads out over an area Earth then re-emits energy at a lower temp Tearth $4\pi D^2$ Must be equal (First Law!) A=πr · Atmosphere is mostly opaque except to Radiation Transmitted by the Atmosphere The Greenhouse Effect visible light & radio waves 0.2 • This gives ~ 278K • ~ 5°C • Note this is an average: is it reasonable? • A bit cool (actually, about 20°C) • Why? Total Absorption • Note the earth is cool, so re-radiated heat is tmosphere at a much lower temp Water Van-Incident energy is (mostly) visible

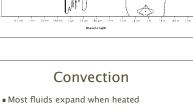
re-radiated is infra-red

Weather

• <u>"Primitive Equations"</u> for weather written down by L F Richardson (1922). Can't be solved without computer

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- Assume we know everything (temperature, pressure, humidity, radiation inflow...) at some points in space.
- Each point will affect it's neighbour, so can figure out how it will change
- Need to know how the energy can be transferred



• Hot water is less dense than cold, so rises



• In atmosphere:

0.2

- ground is heated by sun,
- transfers energy to air
- produces updraft
- in summer, humid air is heated, lifts upwards

Wavelength (um)

Nitrous Oxide Rayleigh Scattering

• cools, water condenses out as cloud

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Evaporation & Condensation

bsorbation solar back to the a

- Evaporation requires a lot of energy
- e.g boiling one litre of water takes 2.3 MJ (million joules)

• Condensation gives the energy back



